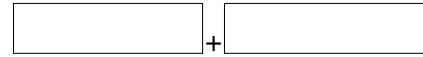
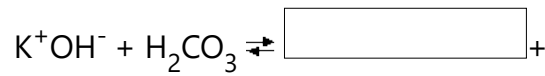
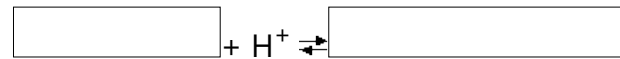
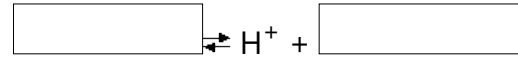
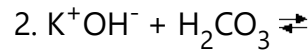
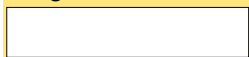
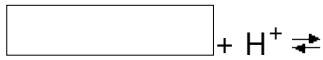
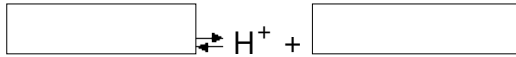
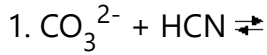


# Exercices sur la réaction acide-base

## I. Exercices : réaction acido-basique :



2. Quels sont, parmi les couples suivant ceux qui ne sont pas des couples acides/bases ?

$\text{C}_6\text{H}_8\text{O}_6 / \text{C}_6\text{H}_7\text{O}_6^-$	$\text{H}_2\text{O} / \text{OH}^-$	$\text{H}^+ / \text{H}_2$	$\text{H}_2\text{SO}_3 / \text{H}_2\text{SO}_4^-$	$\text{NH}_4^+ / \text{NH}_3$
<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>

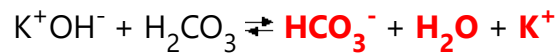
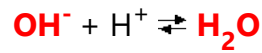
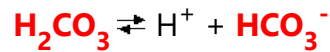
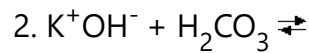
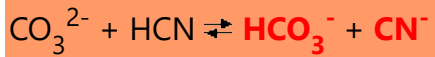
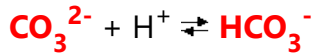
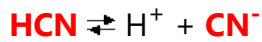
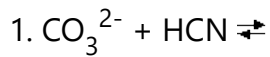
3. Appariez l'acide et sa base conjuguée.

A. $\text{HPO}_4^{2-}$	B. $\text{HCO}_3^-$	C. $\text{HClO}_4$	D. $\text{HCOOH}$	E. $\text{H}_2\text{PO}_4^-$	
1. $\text{HCOO}^-$	2. $\text{HPO}_4^{2-}$	3. $\text{PO}_4^{3-}$	4. $\text{H}_2\text{PO}_4^-$	5. $\text{CO}_3^{2-}$	6. $\text{ClO}_4^-$

réponses :

A.	<input type="text"/>
B.	<input type="text"/>
C.	<input type="text"/>
D.	<input type="text"/>
E.	<input type="text"/>

**II. Correction des exercices : réaction acido-basique :**



2. Quels sont, parmi les couples suivant ceux qui ne sont pas des couples acides/bases ? 3. Appariez l'acide et sa base conjuguée.

$\text{H}_2\text{SO}_3 / \text{H}_2\text{SO}_4^-$	$\text{H}^+ / \text{H}_2$
<input type="checkbox"/>	<input type="checkbox"/>

A.	3
B.	5
C.	6
D.	1
E.	2