

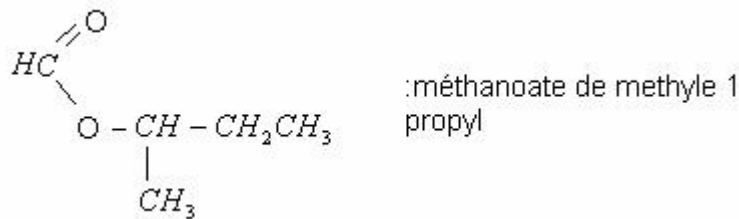
CHIMIE ORGANIQUE.

1° Formule brute de l'ester Ester : $C_nH_{2n}O_2$

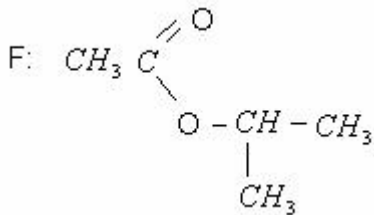
$$M_{\text{ester}} = 102 = 14n + 32 \Rightarrow n = \frac{102 - 32}{14} = 5$$

$C_5H_{10}O_2$

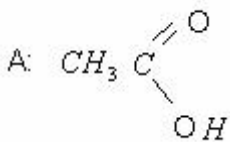
D'où la formule semi-développée de E :



2°



a) Formule semi-développée de A et B :



b) Rendement de cette hydrolyse basique

$$Z = \frac{N_{\text{produit}}}{N_{\text{réactant}}} \times 100$$

$$N_{\text{initial}} = \frac{5,1\text{g}}{102\text{g/mol}} = 0,05\text{mol}$$

$$N_{\text{résidu}} = \frac{2,3\text{g}}{102\text{g/mol}} = 0,0225\text{mol}$$

$$N_{\text{produit}} = 0,05 - 0,0225 = 0,0275\text{mol}$$

$$N_{\text{rendement}} = \frac{0,0275}{0,05} \times 100 = 55\%$$

CHIMIE MINÉRALE :

1) Concentrations molaires des différentes espèces chimiques :

Espèces chimiques, H_2O , H_3O^+ , OH^- , $C_2H_5NH_3^+$, $C_2H_5NH_2$,

$$[H_3O^+] = 10^{-pH} = 10^{-10,8} = 1,58 \cdot 10^{-11} \text{ mol l}^{-1}$$

$$[OH^-] = \frac{10^{-14}}{[H_3O^+]} = \frac{10^{-14}}{1,58 \cdot 10^{-11}} = 0,63 \cdot 10^{-3} \text{ mol l}^{-1}$$

Electroneutralité : $[OH^-] = [C_2H_5NH_3^+] + [H_3O^+]$

$$[H_3O^+] \ll [OH^-] \Rightarrow [OH^-] \approx [C_2H_5NH_3^+] \approx 0,6310^{-3} \text{ mol l}^{-1}$$

$$pK_a = pH - \log \frac{[C_2H_5NH_2]}{[C_2H_5NH_3^+]}$$

$$pK_a = pH - \log \frac{[C_2H_5NH_2]}{[C_2H_5NH_3^+]} \Rightarrow \frac{[C_2H_5NH_2]}{[C_2H_5NH_3^+]} = 10^{-(pK_a - pH)}$$

$$= 10^{0,3}$$

$$= 1,58$$

$$[C_2H_5NH_2] = 1,58 [C_2H_5NH_3^+]$$

$$= 1,58 \times 0,6310^{-3} \text{ mol l}^{-1} = 0,9910^{-3} \text{ mol l}^{-1}$$

2° Valeurs de V_A et V_B

$$V_A + V_B = 30 \text{ ml}$$

$$pK_a = pH - \log \frac{[C_2H_5NH_2]}{[C_2H_5NH_3^+]} = pH - \log \frac{V_B}{V_A}$$

$$\Rightarrow \frac{V_B}{V_A} = 10^{-(pK_a - pH)} = 10^{0,3} = 2,51$$

$$V_B = 2,51 V_A$$

$$V_A + 2,51 V_A = 30$$

$$V_A = \frac{30}{3,51} \text{ mol} = 8,54 \text{ ml}$$

$$V_B = 2,51 \times 8,54 = 21,45 \text{ ml}$$

3° Concentration C'_A de l'acide chlorhydrique :

$$pH = 10,8 = pK_a \Rightarrow [C_2H_5NH_2] = [C_2H_5NH_3^+]$$

$$[Cl^-] = \frac{C'_A V'_A}{V'_B + V_A}$$

$$[C_2H_5NH_3^+] = [Cl^-] = \frac{C'_A V'_A}{V'_B + V_A}$$

Conservation de la matière :

$$[C_2H_5NH_3^+] + [C_2H_5NH_2] = \frac{C'_A V'_B}{V'_B + V_A}$$

$$2[C_2H_5NH_3^+] = \frac{C'_A V'_B}{V'_B + V_A}$$

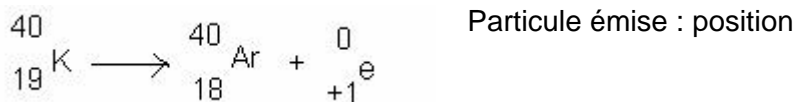
$$[C_1H_5NH_3^+] = \frac{C_B V_B'}{2(V_B + V_A)} + \frac{C_A' V_A'}{V_A + V_B}$$

$$\Rightarrow C_A' = \frac{C_B V_B'}{2V_A'}$$

$$C_A' = \frac{10^{-1} \times 20}{2 \times 20} = 0,510^{-1} \text{ mol l}^{-1}$$

PHYSIQUE NUCLEAIRE

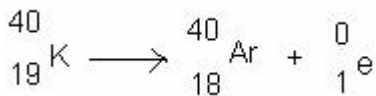
1) Equation de désintégration :



2) Constante radioactive :

$$\lambda = \frac{\ln 2}{T} = \frac{0,69}{1,510^9 \times 365 \times 24 \times 3600} \text{ s}^{-1} = 1,458 \cdot 10^{-17} \text{ s}^{-1}$$

3) age de ces cailloux :



masse du potassium ${}_{19}^{40}\text{K}$ restant $m(t) = 1,66 \cdot 10^{-4} \text{ g}$ nombre de mol

d'Arg on formé :

$$\frac{8210^{-4}}{22,410^3} \text{ mol}$$

$$= 3,66 \cdot 10^{-7} \text{ mol}$$

Nombre de mol de potassium désintégré : $3,66 \cdot 10^{-7} \text{ mol}$

$$\text{Nombre des noyaux de potassium } {}_{19}^{40}\text{K} \text{ restant} = \frac{1,6610^{-4}}{40} \times 6,0210^{23}$$

$$= 0,249 \cdot 10^{17} \text{ noyaux}$$

$$\frac{N_0 - N_0 e^{-\lambda t}}{N_0 e^{-\lambda t}} = \frac{3,6610^{-7} \times 6,0210^{23}}{0,24910^{17}} = 8,848$$

$$(e^{-\lambda t} - 1) = 8,848$$

$$e^{-\lambda t} = 1 + 8,848$$

$$e^{-\lambda t} = 9,848$$

$$t = \frac{\ln 9,848}{\lambda} = \frac{\ln 9,848}{\ln 2} \times T = 4,9710^9 \text{ ans}$$

OPTIQUE GEOMETRIQUE

1° Montrons que $AA' = 4f_1$

$$\gamma = \frac{\overline{OA'}}{\overline{OA}} = -1 \Rightarrow \overline{OA'} = -\overline{OA}$$

$$\overline{AA'} = \overline{AO} + \overline{OA'} = \overline{AO} - \overline{OA}$$

$$\overline{AA'} = -2\overline{OA}$$

$$\frac{1}{f'} = \frac{1}{OA'} - \frac{1}{OA} = -\frac{1}{OA} - \frac{1}{OA} = \frac{-2}{OA} \Rightarrow \overline{OA} = -2f'$$

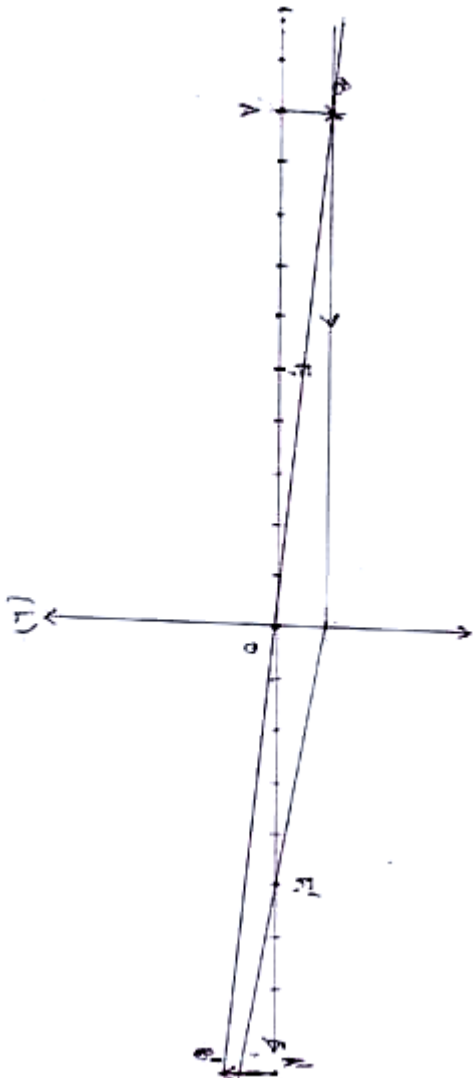
$$\overline{AA'} = -2\overline{OA} = -2(-2f') = 4f'$$

2) Construction géométrique

$$f_1 = \frac{\overline{AA'}}{4} = \frac{1}{4} = 0,25\text{m} = 25\text{cm}$$

$$\overline{OA} = -\frac{\overline{AA'}}{2} = -0,5 = -50\text{cm}$$

Echelle $\frac{1}{5}$



$$\gamma = \frac{\overline{A'B'}}{\overline{AB}} = \frac{\overline{OA'}}{\overline{OA}} = -1 \Rightarrow \overline{OA'} = -\overline{OA}$$

$$\overline{AA'} = 2,5\text{m} = \overline{AO} + \overline{OA'} = \overline{AO} - \overline{OA}$$

$$= -2\overline{OA} =$$

$$\overline{OA} = -\frac{\overline{AA'}}{2} = \frac{-2,5}{2} = -1,25\text{m}$$

$$\frac{1}{f'} = \frac{1}{OA'} - \frac{1}{OA} = -\frac{1}{OA} - \frac{1}{OA} = \frac{-2}{OA}$$

$$f' = -\frac{OA}{2} = -\frac{(-1,25)}{2} = 0,625\text{m}$$

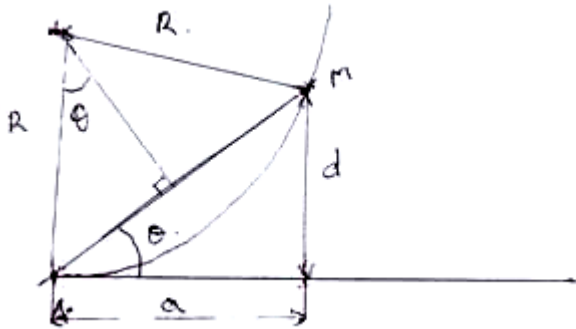
$$C' = \frac{1}{f'} = C_1 + C_2$$

$$\frac{1}{f'} = \frac{1}{f_1} - \frac{1}{f_2} \Rightarrow C_2 = \frac{1}{f'} - C_1 = 1,6 - 4 = -2,4\text{D}$$

$$\boxed{C_2 = -2,4\text{D}}$$

ELECTROMAGNETISME

A 1) Montrons que $R = \frac{d^2 + a^2}{2d}$



$$AM^2 = AO^2 + OM^2 = a^2 + d^2$$

$$\sin \theta = \frac{d}{AM} = \frac{AM}{2R} \Rightarrow AM^2 = 2Rd$$

$$2Rd = a^2 + d^2$$

$$R = \frac{a^2 + d^2}{2d}$$

2) Détermination de charge massique : $\frac{|q|}{m}$

Rayon de la trajectoire : $R = \frac{mV}{eB} = \frac{a_2 + d_2}{2d}$

$$\frac{m}{e} = \frac{a_2 + d_2}{2d} \times \frac{B}{V} \Rightarrow \frac{|q|}{m} = \frac{2dV}{B(a_2 + d_2)}$$

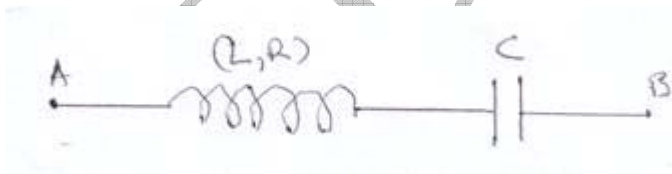
$$\text{D'où } \frac{|q|}{m} = \frac{2 \times 0,1 \times 2 \times 10^7}{0,32(0,5^2 + 0,1^2)} = 4,80 \cdot 10^7 \text{ C/kg}$$

Identification de cette particule :

Électron : $\frac{|q|}{m_e} = \frac{1,610^{-19}}{9,110^{-31}} = 1,7510^{11}$

particule α : $\frac{|q|}{m_\alpha} = \frac{3,210^{-19}}{6,6410^{-27}} = 4,8 \cdot 10^7 \text{ C/kg}$

Donc cette particule est $\alpha = {}^4_2\text{He}$



$$U(t) = U\sqrt{2} \cos(\omega t + \phi)$$

$$u = 100\text{v}$$

$$I = 0,5\text{A}$$

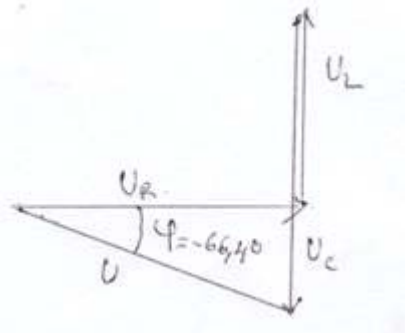
$$U_C = 120\text{V}$$

1° Calcul de Z

$$Z = \frac{U}{I} \quad Z = \frac{100}{0,5} \Omega = 200\Omega$$

$$Z_C \gg U_R$$

2° Calcul de la phase φ



$$\cos \varphi = \frac{U_R}{U} = \frac{RI}{U} = \frac{80 \times 0,5}{100} \approx 0,4$$

$$\cos \varphi = 0,4 \Rightarrow \varphi = 66,42^\circ$$

$$\varphi = -66,42^\circ = 0,369 \pi \text{ rad}$$

$$\boxed{\varphi = -0,369 \pi \text{ rad}}$$

3° Valeur de φ_b

$$U^2 = U_R^2 + (U_C - U_L)^2$$

$$U^2 - U_R^2 = (U_C - U_L)^2$$

$$U_C - U_L = \sqrt{U^2 - U_R^2} = \sqrt{100^2 - 40^2} = 91,65 \text{ V}$$

$$U_L = U_C - 91,65$$

$$U_L = 120 - 91,65 = 28,348 \text{ V}$$

$$\tan \varphi_b = \frac{U_L}{U_R} = \frac{28,348}{40} = 0,708$$

$$\varphi_b = 35,20^\circ = 0,19 \pi \text{ rad}$$

MECANIQUE

A 1° Expression du moment d'inertie J_A

$$J_A = J_D + 2J_S$$

$$J_D = \frac{1}{2} MR^2$$

$$J_S = \frac{2}{5} mR^2 + md^2$$

$$J_A = \frac{1}{2} mR^2 + \left(\frac{2}{5} mR^2 + md^2 \right) 2$$

2) Calcul du moment du couple moteur:

$$\text{TAA. } \sum \mathcal{M}_{A/i_k} = J_A \ddot{\theta}$$

$$\Gamma_m = J_A \ddot{\theta}$$

Calcul de $\ddot{\theta}$:

$$\dot{\theta} = \dot{\theta}_t + \dot{\theta}_0 = \dot{\theta}_t \text{ dont } \dot{\theta}_0 = 0$$

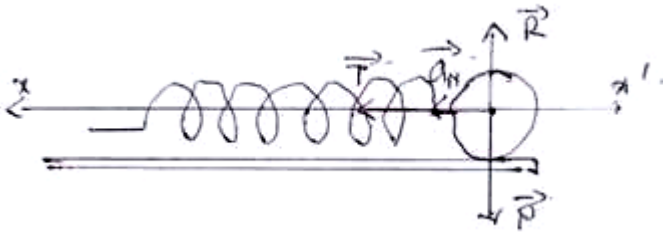
$$\ddot{\theta} = \frac{\dot{\theta}}{t} = \frac{9 \text{ rad s}^{-1}}{4,5 \text{ s}} = 2 \text{ rad s}^{-2}$$

$$\text{D'où } \Gamma_m = J_A \ddot{\theta} \quad J_A = \frac{1}{2} \times 2 \times (0,2)^2 + \left[\frac{2}{5} \times 0,2 \times (0,05)^2 + (0,05)^2 \times 0,2 \right]^2$$

$$J_A = 0,0414 \text{ kg m}^2$$

$$\Gamma_m = 0,0414 \times 2 = 0,0828 \text{ Nm}$$

3° Détermination de la constante raideur du ressort :



TCI $\vec{T} + \vec{R} + \vec{P} = m\vec{a}$

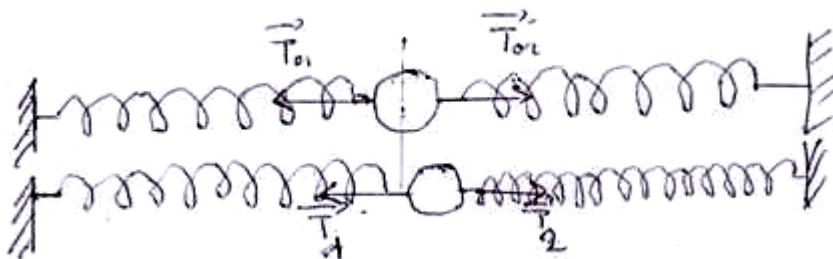
$x'x'$ $T_x + R_x + P_x = ma_H$ dont $P_x = 0$

$$k\Delta l_0 = m \frac{v^2}{R} = m\omega R$$

$$k = \frac{m\omega^2(l_0 + \Delta l_0)}{\Delta l_0}$$

AN $k = \frac{0,2 \times 5^2 (0,08 + 0,02)}{0,02}$

B $k = 25 \text{ N/m}$



1° a) Equation différentielle du mouvement :

Système {sphère + 2ressort} système conservatif

$E_m = \text{constante}$

$E_m = E_c + E_{p_f} + E_{p_{ress}}$

$$E_c = \frac{1}{2} mV^2 = \frac{1}{2} m\dot{x}^2$$

$E_{p_f} = 0$ (Energie potentielle de pesanteur)

$$E_{p_{ress}} = \frac{1}{2} k(\Delta l_0 + x)^2 + \frac{1}{2} k(\Delta l_0 - x)^2$$

$$E_m = \frac{1}{2} m\dot{x}^2 + \frac{1}{2} k(\Delta l_0 + x)^2 + \frac{1}{2} k(\Delta l_0 - x)^2$$

$$\frac{dE_m}{dt} = 0 = \frac{1}{2} m 2V \frac{dV}{dt} + \frac{1}{2} k 2(\Delta l_0 + x) \dot{x} + \frac{1}{2} k 2(\Delta l_0 - x) \dot{x}$$

$$mV\dot{x} + k(\Delta l_0 + x)V - k(\Delta l_0 - x)V = 0$$

$$m\ddot{x} + k\Delta l_0 + kx - k\Delta l_0 + kx = 0$$

$$m\ddot{x} + 2kx = 0$$

$$\ddot{x} + \frac{2k}{m}x = 0 \quad \text{Posons} \quad \omega^2 = \frac{2k}{m} = \frac{2 \times 25}{0,2} = 250$$

$$\omega = 15,81$$

$$\ddot{x} + \omega^2 x = 0$$

C'est une équation différentielle de 2^{nde} ordre à coefficient constant

$$\omega^2 = \frac{2k}{m}$$

b) Equation horaire :

La solution générale est de la forme : $x(t) = x_m \sin(\omega t + \phi)$

$$\text{à } t=0 \quad x(t) = x_m \sin(\omega t + \phi) = x_m = 2 \text{ cm}$$

$$\sin \phi = 1 \Rightarrow \phi = \frac{\pi}{2}$$

$$x(t) = 2 \sin\left(\sqrt{\frac{2k}{m}}t + \frac{\pi}{2}\right)$$

$$x(t) = 2 \sin\left(15,81t + \frac{\pi}{2}\right), \quad x \text{ en cm}$$

2° a) Equation différentielle de mouvement de S

$$\text{T.C.I} \quad \vec{T}_1 + \vec{T}_2 + \vec{P} + \vec{R} + \vec{F} = m\vec{a}$$

$$-k(\Delta l_0 + x) + k(\Delta l_0 - x) - \alpha \dot{x} = m\ddot{x}$$

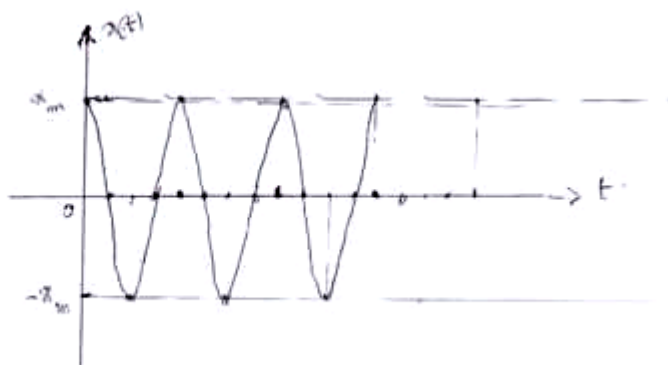
$$-2kx - \alpha \dot{x} = m\ddot{x}$$

$$\ddot{x} + \frac{\alpha}{m} \dot{x} + \frac{2k}{m} x = 0$$

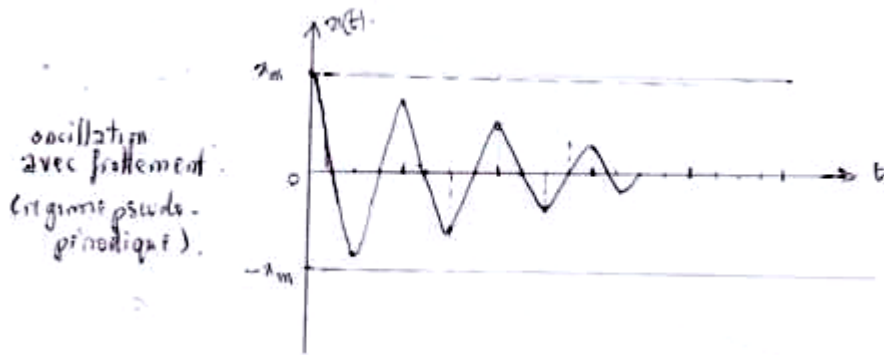
b) Allure de la courbe :

sans frottement

oscillation sans frottement



avec frottement : c'est un régime pseudo périodique



EDUCMAD